

University of Texas Rio Grande Valley

ScholarWorks @ UTRGV

Chemistry Faculty Publications and
Presentations

College of Sciences

5-14-2009

Luminescent Properties of Ensemble and Individual Erbium-Doped Yttrium Oxide Nanotubes

Yuanbing Mao

The University of Texas Rio Grande Valley, yuanbing.mao@utrgv.edu

Xia Guo

Thai Tran

Kang L. Wang

C. Ken Shih

See next page for additional authors

Follow this and additional works at: https://scholarworks.utrgv.edu/chem_fac

 Part of the [Chemistry Commons](#)

Recommended Citation

Mao, Y., Guo, X., Tran, T., Wang, K. L., Shih, C. K., & Chang, J. P. (2009). Luminescent properties of ensemble and individual erbium-doped yttrium oxide nanotubes. *Journal of Applied Physics*, 105(9), 094329. <https://doi.org/10.1063/1.3117520>

This Article is brought to you for free and open access by the College of Sciences at ScholarWorks @ UTRGV. It has been accepted for inclusion in Chemistry Faculty Publications and Presentations by an authorized administrator of ScholarWorks @ UTRGV. For more information, please contact justin.white@utrgv.edu, william.flores01@utrgv.edu.

Authors

Yuanbing Mao, Xia Guo, Thai Tran, Kang L. Wang, C. Ken Shih, and Jane P. Chang

Luminescent properties of ensemble and individual erbium-doped yttrium oxide nanotubes

Cite as: J. Appl. Phys. **105**, 094329 (2009); <https://doi.org/10.1063/1.3117520>

Submitted: 03 October 2008 . Accepted: 17 March 2009 . Published Online: 14 May 2009

Yuanbing Mao, Xia Guo, Thai Tran, Kang L. Wang, C. Ken Shih, and Jane P. Chang



View Online



Export Citation

ARTICLES YOU MAY BE INTERESTED IN

[Controlled erbium incorporation and photoluminescence of Er-doped \$Y_2O_3\$](#)
Applied Physics Letters **87**, 011907 (2005); <https://doi.org/10.1063/1.1984082>

[High-pressure induced phase transitions of \$Y_2O_3\$ and \$Y_2O_3:Eu^{3+}\$](#)
Applied Physics Letters **94**, 061921 (2009); <https://doi.org/10.1063/1.3082082>

[Spectral analysis of synthesized nanocrystalline aggregates of \$Er^{3+}:Y_2O_3\$](#)
Journal of Applied Physics **101**, 113116 (2007); <https://doi.org/10.1063/1.2739316>

Ultra High Performance SDD Detectors



See all our XRF Solutions

Luminescent properties of ensemble and individual erbium-doped yttrium oxide nanotubes

Yuanbing Mao,¹ Xia Guo,² Thai Tran,³ Kang L. Wang,² C. Ken Shih,³ and Jane P. Chang^{1,a)}

¹Department of Chemical and Biomolecular Engineering, UCLA, Los Angeles, California 90095, USA

²Department of Electrical Engineering, UCLA, Los Angeles, California 90095, USA

³Department of Physics, University of Texas, Austin, Texas 78712, USA

(Received 3 October 2008; accepted 17 March 2009; published online 14 May 2009)

The luminescent properties, including cathodoluminescence and photoluminescence, of the erbium-doped yttrium oxide ($\text{Er}^{3+}:\text{Y}_2\text{O}_3$) nanotubes (NTs) have been systematically studied. These NTs were synthesized by a hydrothermal treatment followed by a dehydration process. Cathodoluminescent measurements show that every $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NT is luminescent under electron excitation. In the near-infrared region, sharp, well-resolved, pump-power-dependent, and thermally stable photoluminescence was observed from ensemble NTs. Individual NTs also present characteristic luminescent emissions in the same spectral region. These properties make these NTs promising for applications in display, bioanalysis, and telecommunication.

© 2009 American Institute of Physics. [DOI: 10.1063/1.3117520]

I. INTRODUCTION

Recently, nanosized luminescent materials have attracted intensive attention due to the high surface-to-volume ratio and the quantum confinement effect. The reduction in particle size in a crystalline system can result in remarkable modifications in some of their bulk properties. Therefore, nanosized luminescent materials, i.e., rare-earth (RE)-doped oxides and quantum dots, can exhibit novel physical properties, such as higher luminescent efficiency and better resolution of images in lighting and display.¹⁻³ Moreover, RE-doped oxides possess unique luminescent characteristics deriving from the RE dopants and are very stable in high vacuum. Hence, they have diverse potential applications in nanomaterial-based electronics, photonics, displays, and advanced bioanalyses.^{1,4-8} Because yttrium oxide (Y_2O_3) has the same crystal structure and nearly the same lattice constants with most RE oxides (RE_2O_3), it has received considerable attention as a host material for trivalent RE ions. This similarity in crystal structure allows the incorporation of high concentration of RE ions in Y_2O_3 , compared to that in silicon dioxide. Particularly, numerous studies have been carried out on the $4f^{11}$ Er^{3+} ion, which exhibits the standard telecommunication wavelength emission at $1.54 \mu\text{m}$.⁹

Most of the previous work on RE-doped oxide nanostructures focused on the materials synthesis, morphological and crystal structural characterization, and upconversion mechanism investigation of nanoparticles, nanorods, nanotubes (NTs), and nanowires.¹⁰⁻¹⁵ These nanostructures were synthesized by various procedures, such as the solution-based sol-gel process, gel combustion synthesis, emulsion technique, coprecipitation process, hydrothermal process, template method, electrochemical process, or their combinations.¹⁶⁻³² For example, we have recently reported detailed studies on syntheses and morphological and crystal structural characterization of one dimensional (1D) erbium-

doped yttrium hydroxide [$\text{Er}^{3+}:\text{Y}(\text{OH})_3$] NTs and erbium-doped yttrium oxide ($\text{Er}^{3+}:\text{Y}_2\text{O}_3$) NTs and the effect of local Er^{3+} dopant coordination environments on the photoluminescent spectral features of $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs.^{32,33}

Overall, the optical properties of individual nanosized luminescent emitters have received much attention, mainly on elemental, II-VI, and III-V semiconductors.³⁴⁻³⁸ These low-dimensional systems show interesting fundamental properties and have exciting prospects in nanotechnology-enabled optical and electronic applications.³⁴⁻³⁹ Although RE-doped oxides are important luminescent materials, very few reports investigated the luminescent properties from individual RE-doped oxide nanostructures. One such example is provided by Barnes *et al.*,⁴⁰ who studied on-off blinking and multiple bright states of single europium ions in $\text{Eu}^{3+}:\text{Y}_2\text{O}_3$ nanocrystals (5–15 nm in diameter). These effects have been attributed to the different quasistable Eu^{3+} symmetry sites that modulate the transition of electric dipole moment.

Since RE-doped oxide 1D nanostructures are important building blocks for future optical and optoelectronic nanodevices, it is evident that a thorough understanding of the optical behaviors of a promising material system, such as $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs, can significantly advance the state-of-the-art. Therefore, we report the luminescent properties of $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs in this paper. We first present cathodoluminescent (CL) properties and pump-power-dependent and temperature-dependent photoluminescent behaviors in the near-infrared (NIR) region of ensemble powder samples. Also, photoluminescent behaviors in the NIR region from individual NTs are discussed.

II. EXPERIMENT

A. Sample preparation

The synthesis of $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs is based on a previous report.³² First, $\text{Y}(\text{NO}_3)_3$ and $\text{Er}(\text{NO}_3)_3$ were dissolved in deionized H_2O , followed by the dropwise addition of NaOH

^{a)}Electronic mail: jpchang@ucla.edu.

solution under stirring. After being stirred for 1 h, the solution was transferred into a Teflon-lined autoclave for hydrothermal treatment at 130 °C for 7 h. After being cooled down to room temperature, the precipitate was washed with de-ionized H₂O. Finally, the obtained powder of Er³⁺:Y(OH)₃ NTs was converted into the corresponding powder of Er³⁺:Y₂O₃ NTs by annealing at 500 °C for 3 h in a box furnace.

B. Measurements

In this work, we report optical properties of 5% Er³⁺:Y₂O₃ NTs to show the general trend. CL properties of Er³⁺:Y₂O₃ NTs were analyzed using a CL panchromatic imaging system and spectroscopy attached to a Hitachi S2250-N scanning electron microscope (SEM). Macro-photoluminescence (PL) measurements on ensemble Er³⁺:Y₂O₃ NTs and micro-PL measurements on individual Er³⁺:Y₂O₃ NTs were performed with a 488 nm argon ion laser and a 532 nm frequency doubled diode pumped Nd:yttrium aluminum garnet laser, respectively, both operated with liquid nitrogen cooled InGaAs detectors. For micro-PL measurements, due to the stage and image drift during heating and cooling of the sample stage, it was not possible to track an individual NT and take the measurements at two very different temperatures. While every effort was made to locate a single NT once the stage reached a given temperature, it was not always possible. In this work, an individual NT refers to a single NT or a single NT seemingly clumped/overlapped with no more than one or two much smaller/shorter NTs. To prepare samples for SEM, CL, and PL measurements, powders of Er³⁺:Y₂O₃ NTs were added into ethanol, and the mixtures were subsequently sonicated for about 1 min and later air-dried upon deposition onto silicon wafers.

III. RESULTS AND DISCUSSION

A. SEM and X-ray Diffraction (XRD)

Figure 1(a) shows a typical SEM image of Er³⁺:Y₂O₃ NTs with 5% doping level. The as-prepared NTs are measured to be a few micrometers long and 100–400 nm in outer diameter. As previously reported, the basic 1D topological morphology of the initial Er³⁺:Y(OH)₃ NTs was found to be preserved after a moderate high-temperature annealing process to form Er³⁺:Y₂O₃ NTs.³² This observation suggests that the initial Er³⁺:Y(OH)₃ structural motifs are not affected by the subsequent thermal treatment. In addition, powder and synchrotron XRD analyses confirmed that these 5% Er³⁺:Y₂O₃ NTs are of a cubic bixbyite Y₂O₃ structure (space group: *Ia*3),⁴¹ without observable impurities (data not shown here).^{32,33}

B. CL

By comparing the simultaneously recorded SEM and CL panchromatic images, as shown in Figs. 1(a) and 1(b), it is clear that every Er³⁺:Y₂O₃ NT is luminescent under electron excitation, thereby achieving the observed spatial distribution of luminescence. The CL voltage from these 5%

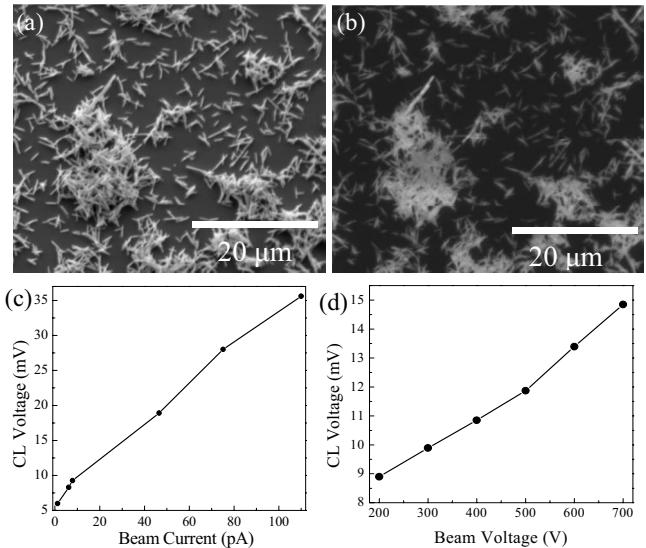


FIG. 1. CL characterization of 5% Er³⁺:Y₂O₃ NTs. (a) SEM image and (b) CL panchromatic image recorded simultaneously under 7 kV accelerating voltage. (c) and (d) are the CL voltage as a function of beam current with a fixed beam voltage of 700 V and that as a function of applied voltage with a fixed beam current of 30 pA, respectively.

Er³⁺:Y₂O₃ NTs gradually increases but does not saturate with increasing electron beam current up to 110 pA under a fixed beam voltage of 700 V [Fig. 1(c)]. The CL voltage increases with increasing acceleration voltages from 200 to 700 V at a fixed beam current of 30 pA [Fig. 1(d)]. Good CL properties are important for developing successful phosphors for field emission display (FED) applications. The excellent CL properties presented here suggest that these Er³⁺:Y₂O₃ NTs are promising luminescent materials for FEDs.

C. PL of ensemble NTs

PL measurements were performed with a 488 nm laser at room temperature on 5% Er³⁺:Y₂O₃ NTs. As shown in Figs. 2(a) and 2(b), these Er³⁺:Y₂O₃ NTs exhibit remarkably sharp and well-resolved peaks at around 870, 980, and 1535 nm, which correspond to the characteristic ⁴S_{3/2} → ⁴I_{13/2}, ⁴I_{11/2} → ⁴I_{15/2}, and ⁴I_{13/2} → ⁴I_{15/2} transitions of Er³⁺ ions, respectively. It indicates that Er³⁺ ions occupy well-defined locations in the Y₂O₃ lattice, i.e., replace the Y³⁺ ions. If Er³⁺ ions have an aperiodic distribution in different crystallographic sites (i.e., interstitial sites), it would randomize the

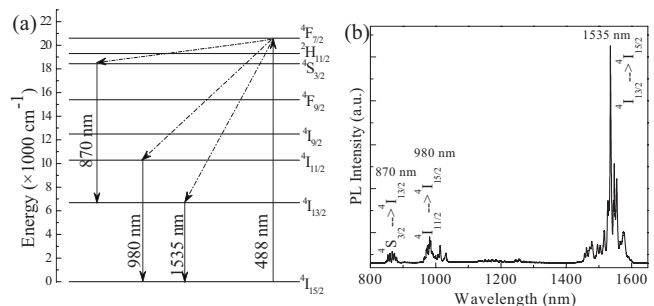


FIG. 2. PL of 5% doped Er³⁺:Y₂O₃ NTs excited at 488 nm at room temperature. (a) An energy level diagram of free Er³⁺ ions. (b) A typical spectrum in the wavelength range of 800 to 1640 nm.

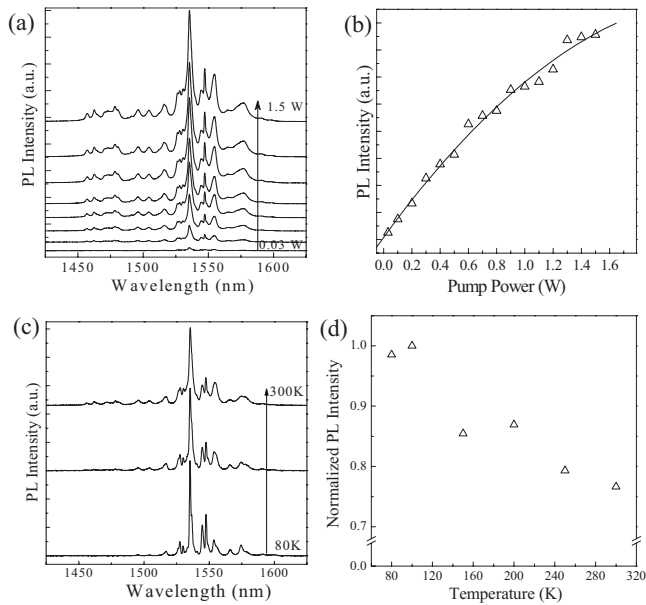


FIG. 3. [(a)–(b)] Pump-power-dependence of 5% doped $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs excited at 488 nm at room temperature. (a) Selected PL spectra (offset for clarification) pumped at various powers from 30 mW to 1.5 W. (b) Integrated PL yield (Δ symbols) in the wavelength range of 1420–1630 nm as a function of pump power. [(c)–(d)] Temperature-dependence of 5% doped $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs excited at 488 nm. (c) Selected PL spectra (offset for clarification) pumped at various temperatures from 80 to 300 K collected at a constant pump power of 0.5 W and a 10 s dwell time. (d) Normalized integrated PL yield (Δ symbols) in the wavelength range of 1420–1630 nm as a function of measurement temperature. The solid line in (b) is used for visual guiding purpose only.

Stark splitting and result in a homogeneous spectral broadening.⁴² These results also indicate that no annealing at temperatures higher than 500 °C is needed to optically activate the erbium ions in these $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs.

In addition, the integrated intensities between 1420 and 1630 nm from PL spectra taken under laser pump powers from 30 mW to 1.5 W [Fig. 3(a)] are plotted in Fig. 3(b) for these $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs. The PL intensity increases with the increasing excitation power, but the peaks exhibit no shift [Fig. 3(a)]. The increase in PL intensity is almost linear as the pump power increases up to about 1.0 W [Fig. 3(b)]. As the pump power increases above 1.0 W, the PL intensity starts to saturate, which represents the onset of a complete excitation. Another notable feature is that a sufficient fraction of Er^{3+} ions are excitable at a pump power as low as 30 mW, as seen from the measurable PL intensity.

We also investigated the temperature-dependent PL quenching of $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs from 80 to 300 K [Figs. 3(c) and 3(d)]. It was reported that the thermal quenching is a major limiting factor for achieving high room-temperature PL yield in Er-doped Si-based materials,^{43–45} with a reduction in PL intensity by at least two orders of magnitude from 200 to 300 K. Our $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs show relatively constant PL intensity with the characteristic Er^{3+} intra- $4f$ transition from 80 to 300 K [Fig. 3(c)]. This lack of thermal quenching is similar to that observed in 50–90 nm $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ thin films synthesized by a low temperature radical-enhanced atomic layer deposition process.⁵ This is attributed to the large band gap of Y_2O_3 (~ 5.6 eV), which inhibits Auger quenching

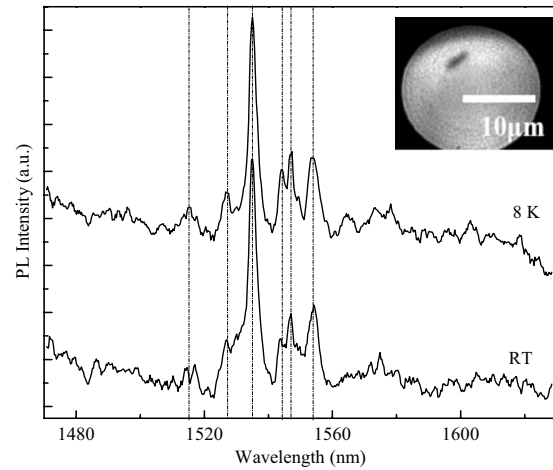


FIG. 4. PL studies of individual 5% $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs excited at 532 nm. PL spectra taken at 8 K and room temperature in the wavelength range of 1470–1630 nm. The inset is the corresponding optical microscopic image of the individual NT for taking the PL spectrum at 8 K.

and energy back-transfer processes, which are found to dominate in Si-based materials.⁴⁶ Also, these $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs do not show any broad luminescent background band, which is characteristic for defect-mediated radiative transitions.^{43–45}

D. PL of individual NTs

The PL properties of individual $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs were investigated by using a 532 nm laser source, while the corresponding morphological images were simultaneously acquired using an optical microscope. Figure 4 shows the characteristic ${}^4I_{13/2} \rightarrow {}^4I_{15/2}$ transitions of Er^{3+} ions in individual $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs at 8 K and room temperature, respectively. The inset of Fig. 4 is a microscopic image of an individual NT, corresponding to the spectrum taken at 8 K. Qualitatively, the spectral line shape and details observed in these individual NTs are very similar to those observed in ensemble NTs by the macro-PL method [Fig. 2(d) versus Fig. 4]. More importantly, individual $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs do not show broad luminescence background bands. These results again validate our previous conclusion that erbium ions occupy well-defined location in the Y_2O_3 lattice, further supported by synchrotron-based XRD, x-ray absorption near-edge spectroscopy, and extended x-ray absorption fine structure studies of $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NT powder samples.^{32,33} Therefore, each of these $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs is a promising luminescent emitter for nanotechnology-enabled optical and electronic applications. Since RE-doped oxide nanostructures have unique and stable luminescent characteristics depending on the specific RE dopants, they are promising alternatives to semiconductor quantum dots.^{34–39} This study on individual $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs thus serves a dual purpose: it sets the stage for a more profound study on optoelectronic properties of individual $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs and paves the way for a similar study on more complex RE-doped oxide nanostructures, e.g., (Er, Yb): Gd_2O_3 nanowires,¹⁰ Eu: Gd_2O_3 nano/microrods,¹⁴ and Eu: GdVO_4 nanorods.¹²

IV. CONCLUSIONS

In summary, the luminescent properties, including cathodoluminescence and both pump-power- and temperature-dependent PL, of ensemble and individual $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs have been systematically studied. The CL measurements show that each $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NT is CL. The luminescent under electron excitation voltage from ensemble NTs gradually increases with both increasing beam current and the increase in the acceleration voltage. These $\text{Er}^{3+}:\text{Y}_2\text{O}_3$ NTs are also highly photoluminescent with remarkably sharp and well-resolved peaks at around 870, 980, and 1535 nm. These NTs show strong pump-power-dependence and, measurable PL intensities with pump power of as low as 30 mW and minor temperature-dependence. Individual NTs also present characteristic luminescent emissions in the NIR region. These results indicate that these materials are promising for applications in display, bioanalysis, and telecommunication.

ACKNOWLEDGMENTS

The authors acknowledge the financial and program support from National Science Foundation (Grant No. CTS0522534), the Office of Naval Research (a Young Investigator Award), the Semiconductor Research Corporation (SRC) and its Focus Center Research Program (FCRP).

- ¹F. Vetrone, J.-C. Boyer, and J. A. Capobianco, in *Encyclopedia of Nanoscience and Nanotechnology*, edited by H. S. Nalwa (American Scientific Publishers, Stevenson Ranch, CA, 2004), Vol. 10, p. 725.
- ²Y. L. Soo, S. W. Huang, Y. H. Kao, V. Chhabra, B. Kulkarni, J. V. D. Veliadis, and R. N. Bhargava, *Appl. Phys. Lett.* **75**, 2464 (1999).
- ³A. Konrad, U. Herr, R. Tidecks, F. Kummer, and K. Samwer, *J. Appl. Phys.* **90**, 3516 (2001).
- ⁴A. J. Steckl and J. M. Zavada, *MRS Bull.* **24**, 16 (1999) (and references therein).
- ⁵T. T. Van, J. Hoang, R. Ostroumov, K. L. Wang, J. R. Bargar, J. Lu, H.-O. Blom, and J. P. Chang, *J. Appl. Phys.* **100**, 073512 (2006).
- ⁶T. Minami, *Solid-State Electron.* **47**, 2237 (2003).
- ⁷J. Yan, M. C. Estevez, J. E. Smith, K. Wang, X. He, L. Wang, and W. Tan, *Nanotoday* **2**, 44 (2007).
- ⁸M. H. Lee, S. G. Oh, S. C. Yi, D. S. Seo, J. P. Hong, C. O. Kim, Y. K. Yoo, and J. S. Yoo, *J. Electrochem. Soc.* **147**, 3139 (2000).
- ⁹A. J. Kenyon, *Prog. Quantum Electron.* **26**, 225 (2002).
- ¹⁰Y. Lei, H. Song, L. Yang, L. Yu, Z. Liu, G. Pan, X. Bai, and L. Fan, *J. Chem. Phys.* **123**, 174710 (2005).
- ¹¹X. Li, Q. Li, J. Wang, and J. Li, *J. Lumin.* **124**, 351 (2007).
- ¹²M. Gu, Q. Liu, S. Mao, D. Mao, and C. Chang, *Cryst. Growth Des.* **8**, 1422 (2008).
- ¹³X. Bai, H. Song, L. Yu, L. Yang, Z. Liu, G. Pan, S. Lu, X. Ren, Y. Lei, and L. Fan, *J. Phys. Chem. B* **109**, 15236 (2005).

- ¹⁴J. Yang, C. Li, Z. Cheng, X. Zhang, Z. Quan, C. Zhang, and J. Lin, *J. Phys. Chem. C* **111**, 18148 (2007).
- ¹⁵X. Bai, H. Song, G. Pan, Y. Lei, T. Wang, X. Ren, S. Lu, B. Dong, Q. Dai, and L. Fan, *J. Phys. Chem. C* **111**, 13611 (2007).
- ¹⁶J. Zhang, S. Wang, T. Rong, and L. Chen, *J. Am. Ceram. Soc.* **87**, 1072 (2004).
- ¹⁷J. A. Capobianco, F. Vetrone, J. C. Boyer, A. Speghini, and M. Bettinelli, *J. Phys. Chem. B* **106**, 1181 (2002).
- ¹⁸H. Eilers, *Mater. Lett.* **60**, 214 (2006).
- ¹⁹G. De, W. Qin, J. Zhang, Y. Wang, C. Cao, and Y. Cui, *J. Lumin.* **119–120**, 258 (2006).
- ²⁰X. Wang and Y. Li, *Chem.-Eur. J.* **9**, 5627 (2003).
- ²¹L. Yang, Y. Tang, X. Chen, Y. Li, and X. Cao, *Mater. Chem. Phys.* **101**, 195 (2007).
- ²²G. S. Wu, Y. Lin, X. Y. Yuan, T. Xie, B. C. Cheng, and L. D. Zhang, *Nanotechnology* **15**, 568 (2004).
- ²³Y.-P. Fang, A.-W. Xu, L.-P. You, R.-Q. Song, J. C. Yu, H.-X. Zhang, Q. Li, and H.-Q. Liu, *Adv. Funct. Mater.* **13**, 955 (2003).
- ²⁴A. M. Pires, O. A. Serra, S. Heer, and H. U. Gudel, *J. Appl. Phys.* **98**, 063529 (2005).
- ²⁵J. A. Capobianco, F. Vetrone, T. D'Alesio, G. Tessari, A. Speghini, and M. Bettinelli, *Phys. Chem. Chem. Phys.* **2**, 3203 (2000).
- ²⁶Z. Xu, Z. Hong, Q. Zhao, L. Peng, and P. Zhang, *J. Rare Earths* **24**, 111 (2006).
- ²⁷Q. Tang, Z. Liu, S. Li, S. Zhang, X. Liu, and Y. Qian, *J. Cryst. Growth* **259**, 208 (2003).
- ²⁸V. V. Rajasekharan and D. A. Buttry, *Chem. Mater.* **18**, 4541 (2006).
- ²⁹T. Hirai, T. Orikoshi, and I. Komasa, *Chem. Mater.* **14**, 3576 (2002).
- ³⁰J. Silver, M. I. Martinez-Rubio, T. G. Ireland, G. R. Fern, and R. Withnall, *J. Phys. Chem. B* **105**, 948 (2001).
- ³¹W. O. Gordon, B. M. Tissue, and J. R. Morris, *J. Phys. Chem. C* **111**, 3233 (2007).
- ³²Y. Mao, J. Huang, R. Ostroumov, K. L. Wang, and J. P. Chang, *J. Phys. Chem. C* **112**, 2278 (2008).
- ³³Y. Mao, J. Bargar, M. Toney, and J. P. Chang, *J. Appl. Phys.* **103**, 094316 (2008).
- ³⁴J. Qi, A. M. Belcher, and J. M. White, *Appl. Phys. Lett.* **82**, 2616 (2003).
- ³⁵D. Kulik, H. Htoon, C. K. Shih, and Y. Li, *J. Appl. Phys.* **95**, 1056 (2004).
- ³⁶J. Wang, M. S. Gudiksen, X. Duan, Y. Cui, and C. M. Lieber, *Science* **293**, 1455 (2001).
- ³⁷J. A. Zapien, Y. Jiang, X. M. Meng, W. Chen, F. C. K. Au, Y. Lifshitz, and S. T. Lee, *Appl. Phys. Lett.* **84**, 1189 (2004).
- ³⁸C. X. Shan, Z. Liu, and S. K. Hark, *Phys. Rev. B* **74**, 153402 (2006).
- ³⁹T. Yatsui, M. Ohtsu, S. J. An, J. Yoo, and G.-C. Yi, *Opt. Rev.* **13**, 218 (2006).
- ⁴⁰M. D. Barnes, A. Mehta, T. Thundat, R. N. Bhargava, V. Chhabra, and B. Kulkarni, *J. Phys. Chem. B* **104**, 6099 (2000).
- ⁴¹JCPDS Card No. 41-1105.
- ⁴²T. T. Van and J. P. Chang, *Appl. Phys. Lett.* **87**, 011907 (2005).
- ⁴³S. Fukatsu, Y. Mera, M. Inoue, K. Maeda, H. Akiyama, and H. Sakaki, *Appl. Phys. Lett.* **68**, 1889 (1996).
- ⁴⁴V. V. Kveder, E. A. Steinman, S. A. Shevchenko, and H. G. Grimmeiss, *Phys. Rev. B* **51**, 10520 (1995).
- ⁴⁵R. Sauer, C. Kisielowski-Kemmerich, and H. Alexander, *Phys. Rev. Lett.* **57**, 1472 (1986).
- ⁴⁶A. J. Kenyon, *Curr. Opin. Solid State Mater. Sci.* **7**, 143 (2003).